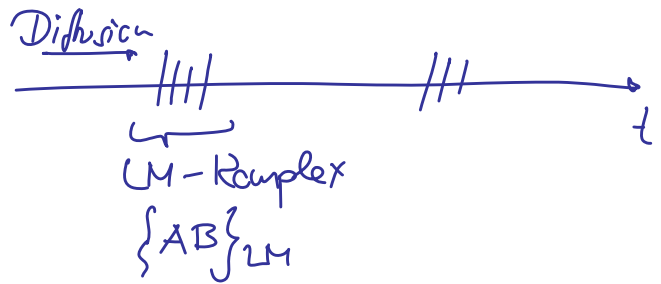
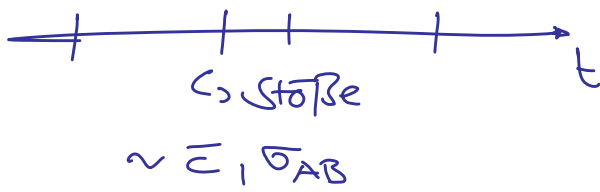


10. Molekulare Reaktionskinetik



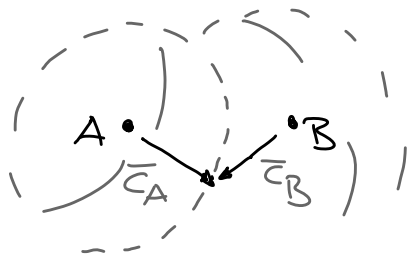
Gasphase

Lösung



A. Stoßtheorie

$$\bar{c} = \sqrt{\frac{8RT}{\pi m}}$$

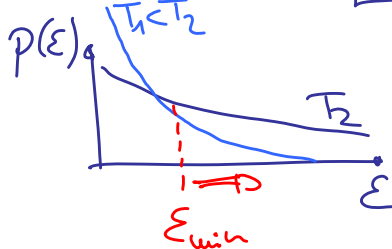


→ Oberflächenintegral

$$\bar{c}_{AB} = \sqrt{2} \cdot \bar{c}$$

Stoßdicke: $Z_{AB} = \bar{c}_{AB} \cdot \sigma_{AB} \cdot \left(\frac{N_A}{V}\right) \left(\frac{N_B}{V}\right) = \bar{c}_{AB} \cdot \sigma_{AB} \cdot N_A^2 \cdot [A][B]$
 $\uparrow = [B] \cdot N_A$

Energie: $\epsilon > \epsilon_{min}$



$$f(\epsilon) = e^{-\epsilon_{min}/RT}$$

$$\left(\frac{1}{L \cdot s}\right)$$

→ molekulare RG:

$$\tilde{v} = f \cdot \bar{c}_{AB} \cdot \sigma_{AB} \cdot N_A^2 [A][B]$$

(mit $1/N_A$)

$$v = \underbrace{\sigma_{AB} \bar{c}_{AB} N_A}_{R_2} \cdot e^{-\epsilon_{min}/RT} [A][B]$$

$\downarrow \bar{c}_{AB} \sim \sqrt{T} \Rightarrow E_A/RT^2 = \frac{\partial}{\partial T} \ln R_2 = \frac{\partial}{\partial T} \left(\ln \sigma_{AB} N_A + \ln \sqrt{2} \sqrt{\frac{8RT}{\pi m}} + \ln T - \frac{\epsilon_{min}}{RT} \right)$
 $= \frac{1}{2T} + \frac{\epsilon_{min}}{RT^2} \Rightarrow E_A = \epsilon_{min} + \frac{1}{2} RT$

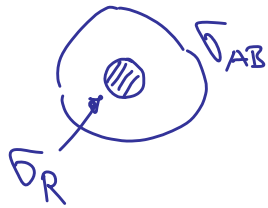
$$\rightarrow R_2 = N_A \sigma_{AB} \bar{c}_{AB} \cdot \sqrt{T} \cdot e^{-E_A/RT}$$

(bei 300K: $RT = 2.5 \text{ kJ/mol}$)

$$\sim 10^{11} \text{ L/mol}\cdot\text{s}$$

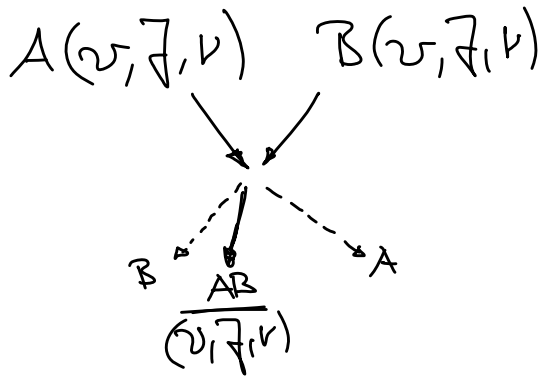
Erweiterung

Reaktionsquerschnitts: $\sigma_R (\sigma^*, \sigma^*(E))$



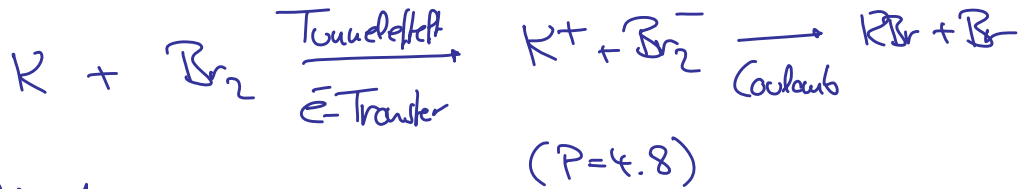
sterischer Faktor $P = \frac{\sigma_R}{\sigma_{AB}}$

Molekularestrahl experimente



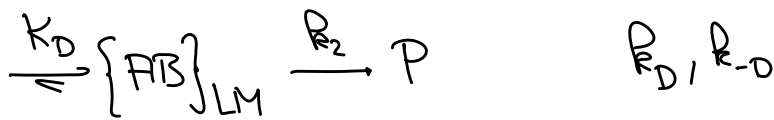
=> Messung $\sigma_R(E)$
(Hinweis: RRKM(M))

Harpoon effekt



=> $P > 1$

Stöße im LM



$$[AB] = \frac{R_D [A][B]}{R_{-D} + R_2}$$

Intermediat

$$\nu_p = R_2 \cdot \frac{R_D [A][B]}{R_{-D} + R_2}$$

$R_2 \gg R_{-D}$

geschw. best.

$R_2 \ll R_{-D}$

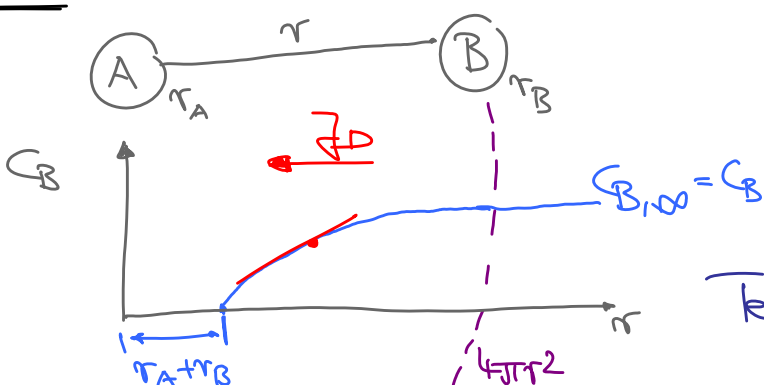
$\nu_p = R_2 \cdot K_D [A][B]$
Reaktionskontrolle

$\nu_p = R_D [A][B]$

Diffusionskontrolle

(Thermodynamik!)

Modell:



Teilchenstrom

$$j_B = \frac{dn_B}{dt} \quad \left(\frac{\text{mol}}{s} \right)$$

Teilchenstrahldichte:

$$j = \frac{j}{A} \quad \left(\frac{\text{mol}}{s \cdot \text{cm}^2} \right)$$

$$j_B/A = + D \frac{dc_B}{dr}$$

Diffusion: $j_D = -D \cdot \frac{dc}{dr} = j/A$

$$\rightarrow j_B = + D_B \cdot \frac{dc_B}{dr} \cdot 4\pi r^2$$

$$j_B \int_{r_A+r_B}^{\infty} \frac{dr}{r^2} = 4\pi D_B \int_0^{\infty} c_B \Rightarrow j_B \cdot \frac{1}{(r_A+r_B)} = 4\pi D_B [B]$$

$$\hookrightarrow j_B = \underbrace{D_B}_{(D_A+D_B)} \cdot 4\pi \underbrace{(r_A+r_B)}_{[A]N_A} [B] \Rightarrow \underbrace{(D_A+D_B)}_{=R_D} \cdot 4\pi \underbrace{(r_A+r_B)N_A}_{[A][B]} = v_p \quad \left[\frac{\text{mol}}{\text{L}\cdot\text{s}} \right]$$

Abschätzung: $D_A = D_B = 10^{-9} \frac{\text{m}^2}{\text{s}}$, $r_A = r_B = 0.5 \text{ nm}$

$$\hookrightarrow R_D = 1.5 \cdot 10^{10} \text{ L/mol}\cdot\text{s}$$

$$D = \frac{RT}{6\pi\eta r} \Rightarrow R_D = 4\pi N_A \left(\frac{RT}{6\pi\eta} \right) \left(\frac{1}{r_A} + \frac{1}{r_B} \right) (r_A + r_B)$$

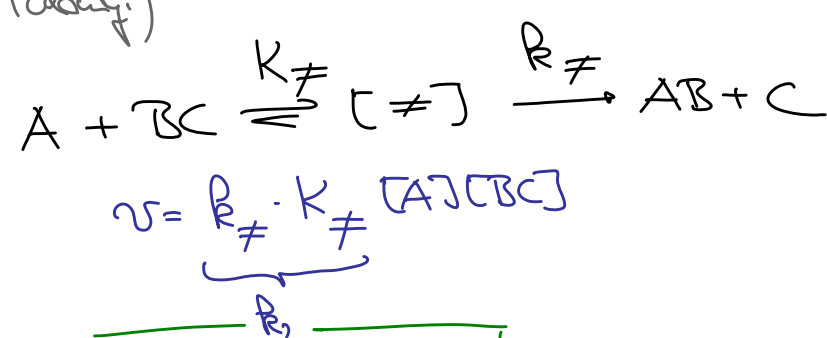
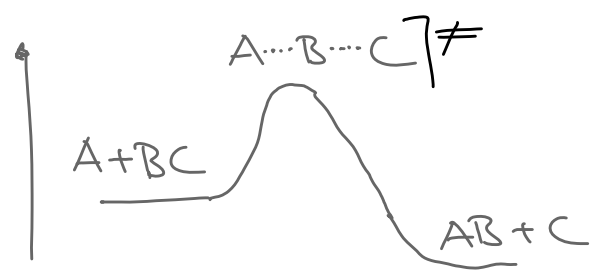
wit $\eta = \eta_0 \cdot e^{\frac{E}{RT}}$
 $E = 15 \frac{\text{kJ}}{\text{mol}}$ in H_2O
 $\frac{(r_A+r_B)^2}{r_A r_B}$

Stoffbilanzgleichung

$$\frac{\partial c_A(t,x)}{\partial t} = D \frac{\partial^2 c_A}{\partial x^2} - v \cdot \frac{\partial c_A}{\partial x} - R \cdot c_A$$

Diffusion Konvektion Reaktion

B. Theorie der UT (Eyring, Polanyi)



$$K_{\neq} = \frac{[A \cdots B \cdots C]}{[A][BC]} = \frac{q_{\neq} \cdot N_A}{q_A \cdot q_{BC}} \cdot e^{-\Delta E_0/RT}$$

↑
statistische TD

"bei \neq ": 1 FG (v_B) wird "aufgegeben"

$$\hookrightarrow R_{\neq} = \left(\frac{RT}{h\nu_{\neq}} \right) \cdot \overline{K}_{\neq}$$

für " \neq " durch
Schwingungs-FG

R_{\neq} : $R_{\neq} = K \cdot \nu_{\neq}$ (K = Transmissionskoeffizient,
i.d.R. $K=1$)

$$\Rightarrow \boxed{R_2 = R_{\neq} \cdot K_{\neq} = K \cdot \frac{RT}{h} \cdot \overline{K}_{\neq}}$$

$6 \cdot 10^{12} \text{ 1/s}$ (\rightarrow vgl. RRKM)

Ergänzung: $\ln \overline{K}_{\neq} = \frac{-\Delta G^{\ddagger}}{RT} = -\left(\frac{\Delta H^{\ddagger}}{RT} - \frac{\Delta S^{\ddagger}}{R} \right)$

$$\Rightarrow \boxed{R_2 = K \frac{RT}{h} \cdot e^{\frac{\Delta S^{\ddagger}}{R}} \cdot e^{-\Delta H^{\ddagger}/RT}}$$

s. sterischer Faktor, $\Delta S^{\ddagger} = \left(\Delta S^{\ddagger}_{A+B \rightarrow AB} + \Delta S^{\ddagger}_{(A+B)} \right)$

(s. Energy-Plots)

C. Potentialhyperflächen

(\rightarrow Folien)